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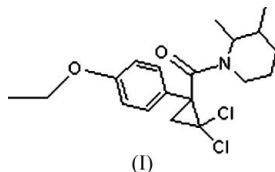
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## Key indicators

Single-crystal X-ray study  
 $T = 298$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.034  
 $wR$  factor = 0.085  
Data-to-parameter ratio = 18.9For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.2,2-Dichloro-1-(4-ethoxyphenyl)cyclopropanyl  
2,3-dimethylpiperidin-1-yl ketoneIn the title compound,  $\text{C}_{19}\text{H}_{25}\text{Cl}_2\text{NO}_2$ , the piperidine ring shows a normal chair conformation and the cyclopropane ring forms a dihedral angle of  $55.68$  ( $17$ ) $^\circ$  with the benzene ring.Received 16 July 2006  
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## Comment

Cycloprothrin derivatives have a high potential for biological activity. They are commonly characterized by low toxicity and good environmental compatibility. As part of our ongoing study on the structure–activity relationships for cycloprothrin derivatives and related compounds, we have recently isolated the title compound, (I), and determined its crystal structure.

The molecular structure of (I) is shown in Fig. 1. The piperidine ring shows a normal chair conformation. The cyclopropane ring forms a dihedral angle of  $55.68$  ( $17$ ) $^\circ$  with the benzene ring. Weak  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonding is observed between neighboring molecules (Table 1).

## Experimental

2,3-Dimethylpiperidine (1.0 g, 8.8 mmol) and triethylamine (1.2 g, 11.9 mmol) were dissolved in dichloromethane (15 ml) with stirring. Then 2,2-dichloro-1-(4-ethoxyphenyl)cyclopropanecarbonyl chloride (2.94 g, 10 mmol) was added dropwise to the mixture at room temperature. The mixture was stirred at room temperature for 15 h, washed three times with water and then dried, yielding 2.93 g of a solid product (yield 90.0%). This was recrystallized from ethanol to give single crystals of (I).

## Crystal data

 $\text{C}_{19}\text{H}_{25}\text{Cl}_2\text{NO}_2$   
 $M_r = 370.32$   
Monoclinic,  $Cc$   
 $a = 11.662$  (5) Å  
 $b = 16.242$  (8) Å  
 $c = 11.398$  (5) Å  
 $\beta = 115.078$  (19) $^\circ$   
 $V = 1955.3$  (15) Å $^3$  $Z = 4$   
 $D_x = 1.258$  Mg m $^{-3}$   
Mo  $K\alpha$  radiation  
 $\mu = 0.34$  mm $^{-1}$   
 $T = 298$  (1) K  
Chunk, colorless  
 $0.33 \times 0.30 \times 0.20$  mm

## Data collection

Rigaku R-AXIS RAPID  
diffractometer  
 $\omega$  scans  
Absorption correction: multi-scan  
(*ABSCOR*; Higashi, 1995)  
 $T_{\min} = 0.888$ ,  $T_{\max} = 0.934$ 9469 measured reflections  
4133 independent reflections  
3347 reflections with  $F^2 > 2\sigma(F^2)$   
 $R_{\text{int}} = 0.024$   
 $\theta_{\text{max}} = 27.5^\circ$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.034$   
 $wR(F^2) = 0.085$   
 $S = 1.02$   
 4133 reflections  
 219 parameters  
 H-atom parameters constrained  
 $w = 1/[0.0004F_o^2 + \sigma(F_o^2)]/(4F_o^2)$   
 $(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.32 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.37 \text{ e } \text{\AA}^{-3}$   
 Extinction correction: Larson  
 (1970), equation 22  
 Extinction coefficient: 145 (16)  
 Absolute structure: Flack (1983),  
 1889 Friedel Pairs  
 Flack parameter: 0.009 (4)

Table 1

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C11-H11\cdots O1^i$	0.93	2.40	3.322 (3)	170

Symmetry code: (i)  $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$ .

Methyl H atoms were placed in calculated positions with  $C-H = 0.96 \text{ \AA}$  and torsion angles were refined to fit the electron density. Other H atoms were placed in calculated positions with  $C-H = 0.93$  (aromatic),  $0.98$  (methine) and  $0.97 \text{ \AA}$  (methylene), and refined in riding mode, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 1996); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *CrystalStructure*.

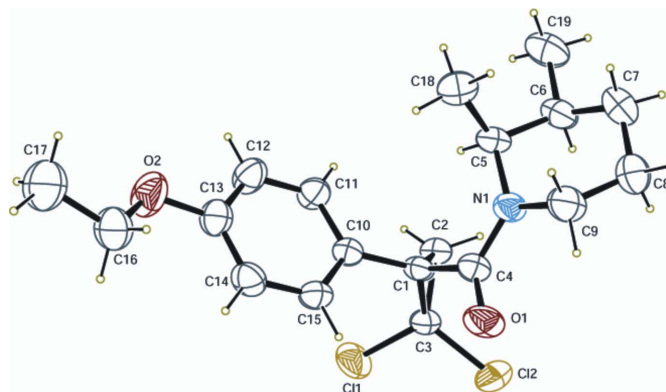


Figure 1

The structure of (I) with 30% probability displacement ellipsoids (arbitrary spheres for H atoms).

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## References

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